CHEM 10 Course Outline as of Fall 2004

CATALOG INFORMATION

Dept and Nbr: CHEM 10 Title: CHEMISTRY & SOCIETY

Full Title: Chemistry and Today's Society

Last Reviewed: 4/30/2007

Units		Course Hours per Week		Nbr of Weeks	Course Hours Total	
Maximum	3.00	Lecture Scheduled	3.00	17.5	Lecture Scheduled	52.50
Minimum	3.00	Lab Scheduled	0	17.5	Lab Scheduled	0
		Contact DHR	0		Contact DHR	0
		Contact Total	3.00		Contact Total	52.50
		Non-contact DHR	0		Non-contact DHR	0

Total Out of Class Hours: 105.00 Total Student Learning Hours: 157.50

Title 5 Category: AA Degree Applicable

Grading: Grade or P/NP

Repeatability: 00 - Two Repeats if Grade was D, F, NC, or NP

Also Listed As:

Formerly:

Catalog Description:

An investigation of the role that chemical concepts play in our modern world. Special emphasis will be given to topics of current interest, including toxic waste disposal and pollution, food additives, drugs, geochemistry, the chemical industry and energy sources.

Prerequisites/Corequisites:

Recommended Preparation:

No science background is necessary. Since extensive reading and writing assignments are required, students must qualify for entry into Engl 1A.

Limits on Enrollment:

Schedule of Classes Information:

Description: An investigation of the role that chemical concepts play in our modern world. (Grade or P/NP)

Prerequisites/Corequisites:

Recommended: No science background is necessary. Since extensive reading and writing assignments are required, students must qualify for entry into Engl 1A.

Limits on Enrollment: Transfer Credit: CSU;UC.

Repeatability: Two Repeats if Grade was D, F, NC, or NP

ARTICULATION, MAJOR, and CERTIFICATION INFORMATION:

AS Degree: Area Effective: Inactive:

C Natural Sciences Fall 2008 Spring 2011
C Natural Sciences Fall 1989 Fall 2004
Transfer Area Effective: Inactive:
B1 Physical Science Fall 2008 Spring 2011

B1 Physical Science Spring 1990 Fall 2004

IGETC: Transfer Area Effective: Inactive: 5A Physical Sciences Fall 2008 Spring 2011

CSU Transfer: Effective: Inactive:

UC Transfer: Effective: Inactive:

CID:

CSU GE:

Certificate/Major Applicable:

Certificate Applicable Course

COURSE CONTENT

Outcomes and Objectives:

A successful student in Chemistry 10 should be able to:

- 1. Show a basic understanding of chemical concepts, including atomic theory, bonding theory, and the kinetic molecular theory.
- 2. Describe the scientific method and how science "works".
- 3. Analyze current problems in modern society as they relate to chemical concepts.
- 4. Describe possible solutions to these current problems.

Topics and Scope:

- 1. Basic Chemical Concepts.
 - a. forms and states of matter
 - b. atomic structure
 - c. chemical bonding
 - d. chemical reactions
 - e. solutions
 - f. energy
- 2. Chemistry and the Environment Within.
 - a. some basic biochemistry
 - b. pharmaceuticals chemistry and disease
 - c. "recreational" drugs the growing scourge
 - d. food and nutrition
 - e. food additives
- 3. Chemistry and the Environment Without.

- a. agricultural chemistry
- b. water and water pollution
- c. air pollution
- 4. Chemistry, Energy, and the "Gaia Concept".
 - a. present sources of energy
 - b. "alternative" energy sources
 - c. earth as an organism

Assignment:

Assignments for Chemistry 10 include:

- 1. specific reading assignments from textbooks as well as from the periodic literature.
- 2. written reports from reading assignments.
- 3. a written term paper.

Methods of Evaluation/Basis of Grade:

Reading reports, Essay exams, Term papers

Writing: Assessment tools that demonstrate writing skills and/or require students to select, organize and explain ideas in writing.

Problem Solving: Assessment tools, other than exams, that demonstrate competence in computational or noncomputational problem solving skills.

Quizzes, Exams

Skill Demonstrations: All skill-based and physical demonstrations used for assessment purposes including skill performance exams.

None

Exams: All forms of formal testing, other than skill performance exams.

Multiple choice, True/false, Matching items, Completion

Other: Includes any assessment tools that do not logically fit into the above categories.

None

Writing 40 - 50%

Problem solving 20 - 30%

Skill Demonstrations 0 - 0%

Exams 20 - 30%

Other Category 0 - 0%

Representative Textbooks and Materials:

CHEMISTRY AND OUR CHANGING WORLD by Sherman and Sherman, Prentice Hall, 1986.

CHEMISTRY FOR CHANGING TIMES by Hill, Burgess Publishing Co., 1986.